

PRODUCT IDENTIFICATION

**Product Name:** Sodium Hydroxide Flakes ACS Grade**CAS Number:** 1310-73-2**Molecular Formula:** HNaO**Molecular Weight:** 39.997 g/mol**Grade:** ACS Grade**Purity / Concentration:** Not Available**Synonyms:** Caustic Soda, Sodium Hydrate

PRODUCT OVERVIEW

Alliance Chemical offers high-purity Sodium Hydroxide Flakes, provided in ACS Grade to ensure rigorous quality standards for sensitive laboratory and industrial processes. With a high assay of 97% and strictly controlled impurity levels—including iron at just 0.9 ppm—these white hygroscopic flakes are ideal for precise pH adjustment and chemical synthesis.

Grade Significance: ACS Grade signifies that this product meets the high-purity standards set by the American Chemical Society, ensuring the reliability and reproducibility required for analytical and laboratory applications.

CERTIFICATE OF ANALYSIS — TYPICAL VALUES

PARAMETER	UNIT	TYPICAL	MIN	MAX	TEST METHOD
Assay (Titration)	%	97	96.5	97.5	Titration
Calcium Ca	ppm	15	0	50	ICP-MS
Iron	ppm	0.9	0	3	ICP
Magnesium Mg	ppm	10	0	20	ICP-MS
Nickel Ni	ppm	5	0	10	ICP-MS
Potassium K	%	0.012	0	0.02	ICP-MS
Chloride (Cl ⁻)	ppm	17	0	100.0	Ion chromatography
Phosphate Po4	ppm	5	0	10	ICP
Sulfate (SO ₄ ²⁻)	ppm	16	0	40	Ion chromatography
Carbonate	%	0.03	0	0.1	Titration
Mercury Hg	ppm	0.3	0	1	ICP-MS
Nitrogen	ppm	5	0	10	Kjeldahl / Chemiluminescence

ND = Not Detected. Values are typical and may vary by lot.

PHYSICAL & CHEMICAL PROPERTIES

Appearance	White hygroscopic flakes	Odor	Odorless
Form	Solid	Boiling Point	greater than 266 °F at 760 mmHg (USCG, 1999)
Melting / Freezing Point	604 °F (USCG, 1999)	Flash Point	Not applicable
Specific Gravity	2.13	Solubility	Highly soluble in water, moderate solubility in alcohols, forms alkaline solutions
Molecular Formula	HNaO	Molecular Weight	39.997 g/mol
Density (25°C)	2.13 g/mL		

APPLICATIONS

1. **Water Treatment** — Used to neutralize acidic water sources and maintain optimal pH levels throughout municipal and industrial treatment processes.
2. **Chemical Manufacturing** — Acts as a powerful base in critical chemical reactions, specifically facilitating efficient saponification and esterification pathways.
3. **Personal Care and Soap** — Serves as an essential reagent in the traditional saponification process required to manufacture high-quality soaps and detergents.
4. **Textile Industry** — Utilized during the dyeing and finishing stages to improve fiber preparation and enhance overall color uptake and vibrancy.

STORAGE & HANDLING

Sodium Hydroxide is highly hygroscopic, meaning it rapidly absorbs moisture from the air, which can cause the flakes to clump and lose their chemical integrity. Proper airtight storage is also critical to mitigate the risk of H290 corrosive reactions with metal containers and to prevent accidental skin or eye contact.

- Store in a cool, dry place away from moisture and incompatible materials.
- Use appropriate personal protective equipment (PPE) including gloves and goggles.
- Ensure containers are tightly sealed to prevent moisture absorption.
- Compatible with HDPE and glass containers; avoid aluminum and zinc materials.
- Provide adequate ventilation in storage areas to minimize exposure to dust.

AVAILABLE PACKAGING

- 2 lbs.
- 4 Lbs.
- 55 Lbs.

SAFETY SUMMARY (CROSS-REFERENCE TO SDS)

Signal Word: Danger



Hazard Statements:

- H290: May be corrosive to metals
- H314: Causes severe skin burns and eye damage

Emergency Contact: CHEMTEL - 800-255-3924 (24 Hours/Day, 7 Days/Week)

For complete safety information, refer to the Safety Data Sheet (SDS) for this product.

Alliance Chemical | 204 South Edmond St, Taylor, Texas 76574 | 512-365-6838 | www.alliancechemical.com

Disclaimer: The information contained herein is believed to be accurate and represents the best information currently available to us. However, Alliance Chemical makes no warranty of merchantability or any other warranty, express or implied, with respect to such information, and we assume no liability resulting from its use. Users should make their own investigations to determine the suitability of the information for their particular purposes.